



Cell Viability Studies of Green Synthesised ZnO Nanoparticles for Antibacterial Properties

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Abstract: The synthesis and characterization of zinc oxide nanoparticles (ZnONP's) using *Ficus religiosa* and *Azadirachta indica* leaves extracts has been presented in this work. Synthesis conditions were optimized for maximal and narrow size range synthesis of zinc oxide nanoparticles. Fine nanorods were formed. The resultant nanomaterials were characterized using various analytical techniques such as SEM, XRD, FTIR, MTT assay. It is suggested that *Ficus religiosa* and *Azadirachta indica* leaf extracts can be effectively used for synthesizing nanoparticles. Also that the green synthesized ZnO nanoparticles can be used as an alternative to existing antimicrobial agents. Nanorods would show good semiconductor properties too.

Keywords: Green Synthesis, Nanoparticles, *Azadirachta Indica*, *Ficus Religiosa*, Characterization, Antimicrobial, Toxicity

1. Introduction

The field of nanomaterials has many applications in science and technology. The synthesis of nanoparticles with specific structure and properties is one of the most important aspects. These studies are extensively done for their magnetic, catalytic activity, electronic, optical, anti-microbial [1], wound healing and anti-inflammatory properties [2]. There are different chemical and physical methods which are more popular for synthesis of nanoparticles. The chemical synthesis methods of ZnO NP's are chemical precipitation, hydrothermal method, pyrolysis, chemical vapour deposition, etc. In the chemical methods, use of toxic compounds and also their residues limits their applications in the field of medicine. Sometimes the synthesis requires high temperature and pressure or even inert atmosphere [3, 4]. On the other hand, 'Green Synthesis Method' for preparation of nanoparticles is efficient and eco-friendly [5-8].

ZnO belongs to metal oxide family with photo-oxidising and photo-catalytic ability against chemical and biological species [9]. ZnO NP's have a wide range of applications such as semi-conductor, piezo electric devices, solar cell, pigment,

catalysis, electronic devices, cosmetic materials, and also remarkable applications in the field of medicine and agriculture. ZnO nanoparticles are known to be one of the adaptable inorganic nanoparticles with efficient anti-bacterial activity [10].

Ficus religiosa belonging to Moraceae family, commonly known as Peepal (Ravi) tree is a medicinally important tree species. It is considered a sacred tree in India and is respected by followers of many religions. It is extremely popular in indigenous system of medicine like Ayurveda, Siddha, Unani and Homeopathy. Studies have been carried out in the past that validate the antimicrobial property of *Ficus religiosa* and have been documented and this tree is still regarded as "Village dispensary" in India [11]. Phytochemical investigation of this plant barks show the presence tannins, saponins, flavonoids, steroids, terpenoids and cardiac glycosides. Each part of the *Azadirachta indica* plant which is commonly available in India has been used as a household remedy against various human ailments from antiquity and for treatment against viral, bacterial and fungal infections [10]. Several researchers have submitted articles on green

synthesis of ZnO [12-18] and other nanoparticles, but each material synthesised has its own significance in terms of experimental conditions and resultant size of the particles. Thus this study focus on the easy synthesis of ZnO nanoparticles and study of the morphology, structure and stability of ZnO using Spectroscopic techniques like: SEM, XRD, FTIR and cytotoxicity studies.

2. Materials and Methods

2.1. Preparation of Leaf Extract

Fresh leaves of *Ficus religiosa* and *Azadirachta indica* were collected and washed 4 times with distilled water to remove the dust particles. Wetness was removed from them by placing them on blotting paper. The leaves were microwave dried separately for 4 minutes for *Ficus religiosa* and 3 minutes for *Azadirachta indica*. Care was taken to see that they remained green in colour. Then 10g of smashed leaves, were added to 100 ml of distilled water and heated at 60°C for 30 minutes until the colour of aqueous solution changes from transparent to greenish colour. Then the extract was cooled to room temperature and filtered using filter paper and used for further process.

2.2. Preparation of ZnONP's

i) 50 ml *Ficus religiosa* leaf extract was taken in a glass beaker and 5gm ZnNO₃ was added to it and the solution was heated on a hot plate between 70°C – 80°C for 55 min. When the solution reduced to deep reddish paste, it was transferred into a crucible and kept in hot air oven at 150°C for 45 min. Whitish-yellow soft powder was obtained without any further chemical treatment.

ii) 100 ml of *Azadirachta indica* leaf extract was taken in a beaker and 5g of zinc nitrate was added and heated under continuous manual stirring for 60 minutes, the temperature was maintained 60 – 70°C. It reduces brownish-yellow paste. Then the paste is dried in the hot air oven at a temperature of 100 degrees for 60 min. The paste turned into fine whitish yellow powder and is further used for characterisation purposes.

2.3. Cell Lines and Culture

Standard murine macrophage cell line (RAW264.7A) procured from ATCC and human epithelial adenocarcinoma (HeLa) cell lines were used and cultured in RPMI-1640 medium containing 2 mM L-glutamine supplemented with 10% FCS, 100 iU/mL penicillin and 100 µg mL⁻¹ streptomycin. Cells were cultured at 37°C in a humidified incubator at 5% CO₂.

2.4. MTT Assay

Mitochondrial/metabolic activities of RAW macrophages and HeLa cells in response to ZnONP's was investigated by Tetrazolium salt, 3-(4,5-dimethylthiazol-2-yl)-3,5-diphenyltetrazolium (MTT) assay. Cells were

seeded overnight for adhesion. The next day, cells were washed and cultured with different concentrations of polymer for 24 h. Following incubation cells were washed and incubated with MTT reagent for 2 h and formazan crystals were dissolved in dimethyl sulfoxide (DMSO) and the absorbance was read at 570 nm with reference at 630 nm using multimode plate reader (TECAN; Infinite M200PRO). The percentage of cell viability was calculated.

3. Results and Discussions

3.1. Scanning Electron Microscopy (SEM)

The Scanning Electron Microscopy (SEM) provides further insight into the morphology and size details of the ZnO nanoparticles. The SEM images of ZnONP's are shown in the figures 1 and 2. It revealed the information about formation of crystals in the form of nano-rods and also in nanoscale.

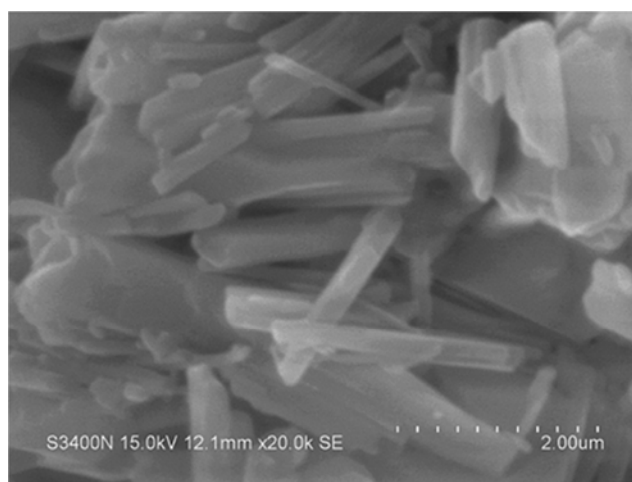


Figure 1. SEM image of ZnO prepared using *Ficus religiosa*.

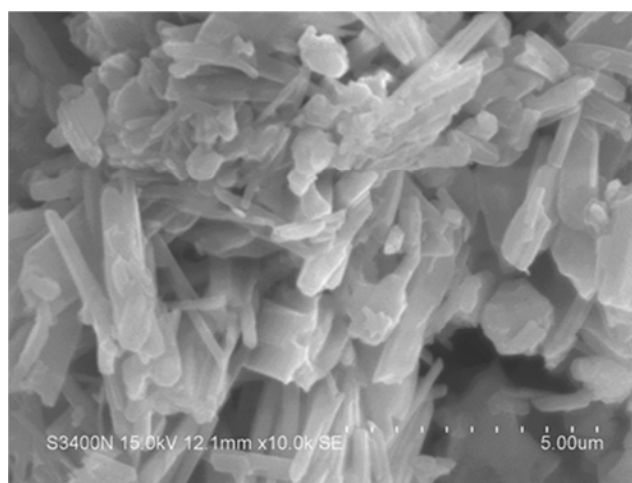


Figure 2. SEM image of ZnO prepared using *Azadirachta indica*.

3.2. X-Ray Diffractometry (XRD)

The XRD pattern of ZnO nanoparticles is shown in figures

3 and 4. The observed diffraction peaks at 31.45°, 33.95°, 37.50°, 47.10°, 56.08°, 62.51°, 67.58° which correspond to pure ZnO. The peaks of graph are in good match with the literature report [7]. The formations of ZnO particles was confirmed. The average size of particle was calculated by using Scherrer's formula. Using *Ficus religiosa*, the average particle size was 23nm and that of *Azadirachta indica* was 25nm.

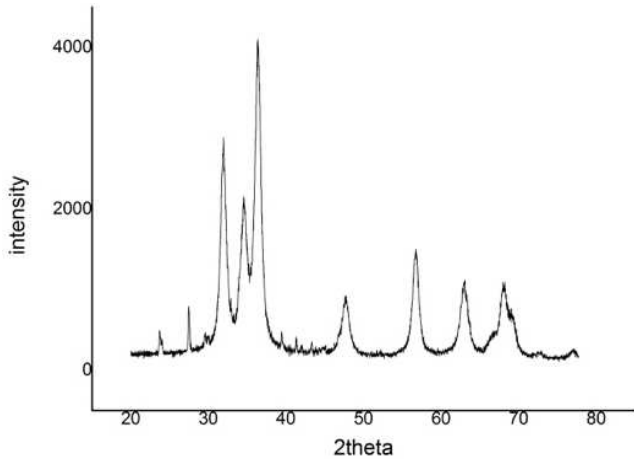


Figure 3. XRD of ZnO synthesised using *Ficus religiosa*.

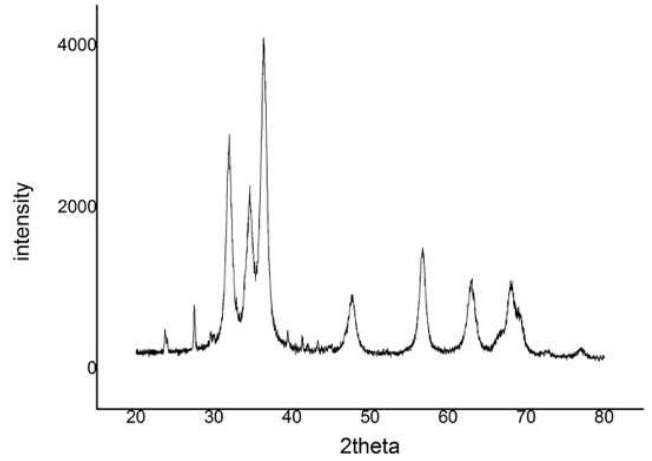


Figure 4. XRD of ZnO synthesised using *Azadirachta indica*.

3.3. Fourier Transform Infra-Red (FTIR) Spectroscopy

The sample peak observed between the band position range 1000 - 1300 cm^{-1} in figures 5 and 6 corresponds to C-O stretch and 1300 - 1400 cm^{-1} corresponds to N=O stretch. The band at 1625 cm^{-1} - 1750 cm^{-1} attribute to C=O stretch indicating the compound to be ketones. The Broad peak at 31000 cm^{-1} - 3600 cm^{-1} corresponds to O-H and C=O indicating the compound to be alcohols and carboxylic acid. The peaks at 790 cm^{-1} - 840 cm^{-1} correspond to CH bend.

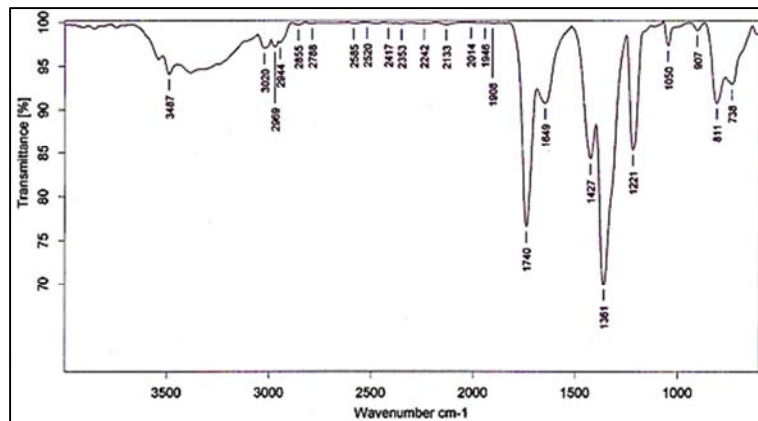


Figure 5. FTIR of ZnO synthesised using *Ficus religiosa*.

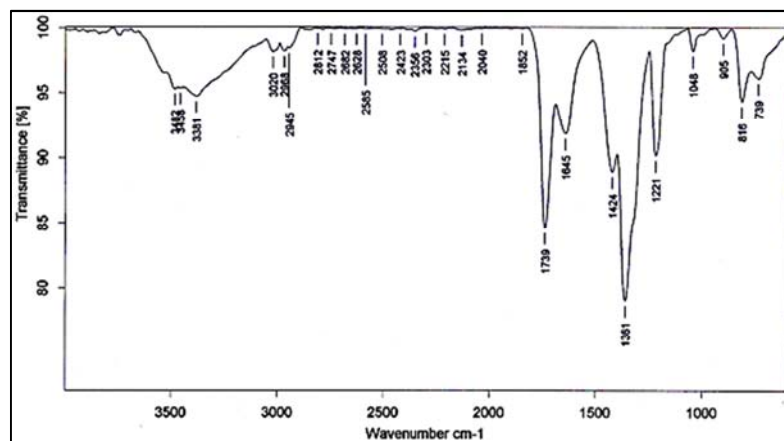


Figure 6. FTIR of ZnO synthesised using *Azadirachta indica*.

3.4. MTT Assay

After synthesis, we tested the influences of these nanoparticles on cell viability. To that purpose, we checked cytotoxicity associated with ZnO NP's on both macrophages (Figure 7) and epithelial cells (Figure 8). The ZnO NP's remained safe in both cell lines tested at the concentration 125 μ g/ml tested showing ~75% cell viability. Interestingly, ZnO NP's capsules at 62.5 μ g/ml showed no sign of toxicity (Figure 7) in both the cell lines tested.

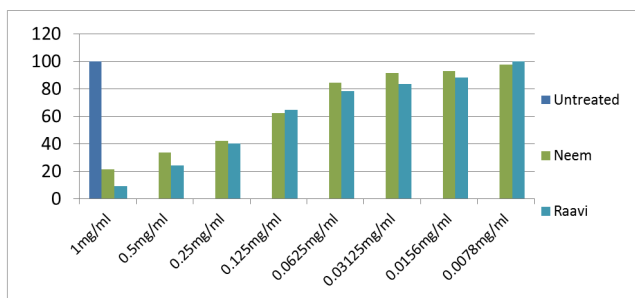


Figure 7. MTT Assay of ZnO@Ravi/Neem synthesised using macrophages.

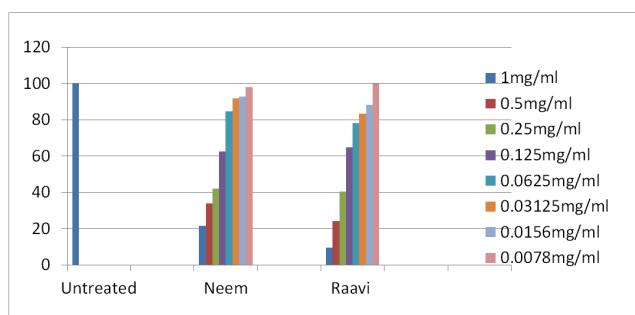


Figure 8. MTT Assay of ZnO@Ravi/Neem using epithelial cells.

3.5. Anti-Bacterial Studies

Laboratory grade ZnO, ZnO synthesised using Ficus Religiosa extract and Azadirachta Indica were tested for antibacterial properties. Two bacteria, namely: E. Coli and Streptococcus aureus were used for straining on agar plates and the inhibitions were observed. Functionalized ZnO because of the leaf extracts showed inhibitions of bacterial growth compared to the laboratory grade ZnO when tested with the two bacteria.

4. Conclusions

SEM images confirm the formation of nanosized ZnO rods, whereas XRD graphs confirm ZnO crystal structure formation along with the particle size in nano dimensions. Functionalization of nanorods because of Ficus Religiosa and Azadirachta Indica leaf extracts can be clearly visualized in FTIR plots. Zinc Oxide nanorods synthesized using Azadirachta indica and Ficus religiosa leaf extracts did not require any further processing or grinding as in the case of other synthesis methods. The furnace temperatures used were very low (100-150 $^{\circ}$ C) compared to sol-gel synthesis. Green

synthesized Zinc Oxide nanoparticles in the present study show perfect cell viability and good antibacterial activity. It was found that Azadirachta indica and Ficus Religiosa leaves extract are susceptible to the bacterial strains. It is a cheaper and efficient method for obtaining functionalized nanomaterials. The method can be adopted for several other metal and metal oxide synthesis and can be extended to other plant extracts; there is a larger scope of research because of the availability of abundant biodiversity.

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